Differential Roles of Human Monoamine (M)-Form and Simple Phenol (P)-Form Phenol Sulfotransferases in Drug Metabolism

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Cytosolic sulfotransferases (STs) are traditionally known as Phase II drug-metabolizing or detoxifying enzymes that facilitate the removal of drugs and other xenobiotic compounds. In this study, we carried out a systematic investigation on the sulfation of drug compounds by two major human phenol STs (PSTs), the monoamine (M)-form and simple phenol (P)-form PSTs. Activity data obtained showed the differential substrate specificity of the two enzymes for the thirteen drug compounds tested. Kinetic studies revealed that the M-form PST displayed stereoselectivity for the chiral drug, isoproterenol. The effects of divalent metal cations on the activity of the M-form and P-form PSTs toward representative drug compounds were quantitatively evaluated. Results obtained indicated that the drug-sulfating activities of the two human PSTs were partially or completely inhibited or stimulated by the ten divalent metal cations tested at a 5 mM concentration. The two enzymes appeared to be less sensitive to the effects of physiologically more abundant metal cations such as Mg^{2+} and Ca^{2+} , but more sensitive to the detrimental effects of other metal cations that may enter the body as environmental contaminants.

Key words: drug metabolism, phenol sulfotransferase, stereoselectivity, substrate specificity, sulfation.

Abbreviations: ST, sulfotransferase; PST, phenol sulfotransferase; PAPS, 3'-phosphoadenosine 5' phosphosulfate.

Sulfate conjugation is a major pathway *in vivo* for the biotransformation and excretion of drugs and xenobiotics, as well as endogenous compounds such as steroid and thyroid hormones, catecholamines, cholesterol, and bile acids (1-3). The responsible enzymes, called the "cytosolic sulfotransferases (STs)", catalyze the transfer of a sulfonate group from the active sulfate, 3'-phosphoadenosine 5'-phosphosulfate (PAPS), to an acceptor substrate compound containing either a hydroxyl or an amine group (4). Sulfate conjugation may result in the inactivation or activation of the substrate compounds or increase their water-solubility, thereby facilitating their removal from the body (1-3).

In searching for the cytosolic ST enzymes involved in the sulfation of drugs, earlier studies revealed two distinct forms of the "phenol ST" (PST) in human platelets (5, 6). The monoamine (M) (thermolabile)-form PST catalyzes more effectively the sulfation of monoamines such as dopamine and epinephrine, and the simple phenol (P) (thermostable)-form PST preferentially sulfates neutral phenols such as p-nitrophenol and α -naphthol (7). It is now clear that, in addition to their presence in platelets, these two PSTs have a somewhat widespread tissue distribution in the human body. In particular, the M-form PST has been found in the upper gastro-intestinal tract

and brain (8), and the P-form PST in the adrenal gland, lung, and liver (9). Both forms are thought to be constitutive enzymes though little is known about the regulation of their enzymatic activity (1). In the past several years, however, studies performed in our laboratory have revealed that some divalent metal cations may exert stimulatory or inhibitory effects on these enzymes (10, 11). For example, the addition of Mn^{2+} to the reaction mixture resulted in a dramatic increase in the Dopa/tyrosine-sulfating activity of the M-form PST, while Cd²⁺ inhibited this activity. These findings indicate that divalent metal cations may play a significant role in regulating the activity of the M-form PST and perhaps cytosolic STs in general. It may therefore be important to investigate whether divalent metal cations that enter the body in food or as environmental contaminants exert significant stimulatory or inhibitory effects on the sulfation of drugs. Another interesting aspect of the metabolism of drugs through sulfation is the stereoselective sulfate conjugation of some of these compounds. A study using rat hydroxysteroid ST (STa) has revealed the stereoselective sulfation of chiral secondary alcohols (12). Studies using homogenates of, or intact, human liver, intestine, or platelets have also demonstrated the stereoselective sulfation of some chiral drugs including isoproterenol and albuterol (13, 14). In view of our previous studies showing that the M-form PST displayed stereoselectivity for the D-enantiomers of Dopa and tyrosine (11), it is interesting to investigate whether the M-form PST also exhibits stereoselectivity toward chiral drugs.

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We report here a systematic investigation on the differential activities of the M-form and P-form PSTs toward a variety of drug compounds. The kinetics of sulfation of the two enantiomeric forms of the chiral drug isoproterenol by the M-form PST was examined. Moreover, the effects of divalent metal cations on the drug-sulfating activities of these two enzymes were studied.

MATERIALS AND METHODS

Materials—Acebutolol, acetaminophen, adenosine 3'phosphate 5'-phosphosulfate (PAPS), bupivacaine, dexamethasone, dobutamine, lidocaine, (±)isoproterenol, (+)isoproterenol, (–)isoproterenol, 4-methylumbelliferone, (±)metoprolol, minoxidil, DL-propanolol, (±)salbutamol, (±)sotalol, dopamine, p-nitrophenol, and adenosine 5'triphosphate (ATP) were products of Sigma. Carrier-free sodium [³⁵S]sulfate was from ICN Biomedicals. Cellulose thin-layer chromatography (TLC) plates were from EM Science. All other chemicals were of the highest grade commercially available.

Preparation of Purified Human M-Form and P-Form PSTs—Human M-form (SULT1A3) and P-form (SULT1A1) PSTs were cloned, expressed, and purified using the pGEX-2TK Glutathione S-transferase Gene Fusion System based on the procedure previously established (15, 16).

Enzymatic Assay-Sulfotransferase activities of purified M-form and P-form PSTs were assayed using [³⁵S]PAPS as the sulfonate group donor. The standard assay mixture, in a final volume of 25 µl, contained 50 mM potassium phosphate buffer (pH 7.0), 15 µM $[^{35}S]$ PAPS, and 50 μ M of the drug compound tested. The enzyme dilutions were prepared in 50 mM potassium phosphate buffer (pH 7.0) containing 10% glycerol and 8 mM dithiothreitol. The reaction was started by the addition of 5 µl of the enzyme preparation, allowed to proceed for 3 min (so that the reaction reached no more than 5-10% of completion) at 37°C, and terminated by heating at 100°C for 2 min. The precipitates were cleared by centrifugation for 1 min, and the supernatant was subjected to the analysis of [35S]sulfated product based on the TLC procedure previously established (17). Each experiment was performed in triplicate, together with a control without enzyme. The results obtained were calculated and expressed in nanomoles sulfated product formed/min/mg protein.

Determination of the Stimulatory/Inhibitory Effects of Divalent Metal Cations on the Sulfation of Drug Compounds by Human M-Form or P-Form PST-To determine the patterns of stimulation/inhibition of purified human M-form or P-form PST by divalent metal cations, enzymatic assays were performed in the presence or absence of such cations. Standard assay mixture, in a final volume of 25 µl, contained the test divalent cation (5 mM), 14 µM PAP^{[35}S]. 50 mM potassium phosphate buffer (pH 7.0), the enzyme being assayed, and a specified concentration of the drug compound as substrate. Controls containing all the reagents, but without divalent metal cation or with 5 mM EDTA, were assayed in parallel. The reaction was started by the addition of the enzyme and allowed to proceed for 3 min at 37°C, followed by heat inactivation of the enzyme at 100°C for 2 min. The precipitates were cleared by centrifugation for 1 min, and

Table 1. Specific activity of human M-form and P-form PSTs.ª

$Drug \ compounds \ (50 \ \mu M)$	Specific activity (nmol/min/mg protein)		
	M-PST	P-PST	
Acebutolol	N.D. ^b	2.9 ± 0.2	
DL-Propanolol	N.D.	2.4 ± 0.1	
Bupivacaine	N.D.	2.0 ± 0.1	
(±) Metoprolol	N.D.	2.0 ± 0.1	
(±) Sotalol	N.D.	2.3 ± 0.1	
Dexamethasone	N.D.	2.7 ± 0.1	
Dobutamine	68.8 ± 4.4	106.0 ± 2.0	
4-Methylumbelliferone	9.4 ± 0.6	95.1 ± 2.9	
Lidocaine	0.05 ± 0.02	0.3 ± 0.1	
(\pm) Isoproterenol	147.0 ± 1.1	22.1 ± 1.1	
(\pm) Salbutamol	25.5 ± 0.2	2.9 ± 0.5	
Acetaminophen	2.8 ± 0.1	11.7 ± 0.8	
Minoxidil	0.5 ± 0.1	4.8 ± 0.1	

^aData shown represent the mean \pm SD derived from three determinations. ^bActivity not detected.

the supernatant was subjected to the analysis of $[^{35}S]$ sulfated product based on the TLC procedure previously established (17).

Miscellaneous Methods—[³⁵S]PAPS was synthesized from ATP and carrier-free [³⁵S]sulfate using the bifunctional human ATP sulfurylase/APS kinase and its purity determined as previously described (18). The [³⁵S]PAPS synthesized was then adjusted to the required concentration and specific activity by the addition of cold PAPS. SDS–polyacrylamide gel electrophoresis was performed on 12% polyacrylamide gels using the method of Laemmli (19). Protein determination was based on the method of Bradford with bovine serum albumin as standard (20).

RESULTS AND DISCUSSION

Biological sulfation was first discovered when phenyl sulfate was isolated from the urine of a patient who had been treated with phenol (21). This finding had largely positioned the research on sulfation and the responsible sulfotransferase enzymes within the general area of pharmacology/toxicology for over a century. Indeed, a great many studies using experimental animals or volunteer human subjects have demonstrated the metabolism of drugs through sulfation (22, 23). The current study aimed to examine the differential roles of the Mform and P-form PSTs in the sulfation of drug compounds and the regulatory effects of divalent metal cations.

Differential Activities of Human M-Form and P-Form PSTs toward Drug Compounds—Purified M-form and Pform PSTs were assayed using a variety of drug compounds as substrates. Activity data compiled in Table 1 showed that all drug compounds tested (at a concentration of 50 μ M in the assay mixture) could be used as substrates by the P-form PST, with dobutamine and 4-methylumbelliferone giving the highest specific activities. In contrast, the M-form PST displayed a more distinct substrate preference, with high activities toward dobutamine, salbutamol, and (±)isoproterenol (all of which are dopamine analogs) and very low or undetectable activities with the other substrates (the concentration of the substrates used was 50 μ M). These results showed clearly the differential substrate specificity of the M-form and P-form PSTs toward the drug compounds tested. It is interesting to note that, in line with its general role as a detoxifying enzyme (2, 3), the P-form PST displays a broader substrate specificity than the M-form PST, which is believed to play a more specific role in the homeostasis of dopamine in the brain and perhaps the detoxification of this and other deleterious monoamines and dopamine analogs in the upper gastrointestinal tract (8). Table 2 shows the kinetic constants determined for the sulfation of acetaminophen (one of the most extensively used drugs, which is believed to be detoxified in the liver) and minoxidil (which is activated by conversion to its sulfateconjugated form) by the P-form PST and the sulfation of dobutamine by the M-form PST. Based on these data, the catalytic efficiency of the M-form PST for the sulfation of dobutamine appeared to be considerably higher than that of the P-form PST for the sulfation of acetaminophen or minoxidil. Previous studies using liver homogenates demonstrated the stereoselective sulfation of (\pm) isoproterenol (13). In view of our previous finding that the Mform PST displayed stereoselectivity for the D-enantiomers of Dopa and tyrosine (11), we decided to test the kinetics of sulfation of the two enantiomeric forms of isoproterenol by this enzyme. Results shown in Fig. 1 indicated that the apparent $K_{\rm m}$ values determined for the two enantiomers of isoproterenol are nearly identical, whereas the $V_{\rm max}$ for the (+)enantiomer is almost two times that for the (-)enantiomer. The preferential sulfation of one enantiomeric form over the other may have important pharmacological implications for drugs that are administered in the racemic form.

Effects of Divalent Metal Cations on the Sulfation of Drug Compounds by Human M-Form and P-Form PSTs—Divalent metal cations are known to play important roles in the function of biological molecules (24). Some of them, including manganese, zinc, copper, iron, and cobalt, are essential components of different enzymes, while others such as magnesium and calcium are required for the actions of certain enzymes or proteins. In contrast to these biologically useful metal cations, some metal cations (e.g., lead and mercury) that enter the body primarily as environmental contaminants have been shown to exert deleterious effects (25, 26). Our

Table 2. Kinetic constants of human M-form and P-form PSTs with dobutamine, acetaminophen, and minoxidil as substrates.^a

	$K_{\rm m}(\mu{ m M})$	V _{max} (nmol/min/mg)	$V_{\rm max}/K_{\rm m}$
M-form PST			
Dobutamine	14.8	384.6	26.0
P-form PST			
Acetaminophen	430.4	22.1	0.05
Minoxidil	200.6	6.2	0.03

^aData shown represent mean \pm SD derived from three determinations.

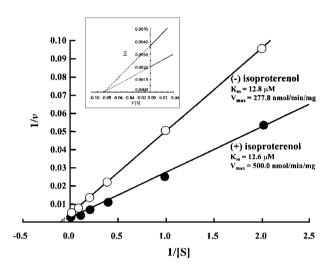


Fig. 1. Lineweaver-Burk double-reciprocal plot of the human M-form PST with (+)Isoproterenol or (-)Isoproterenol as substrate. Concentrations of (+)isoproterenol or (-)isoproterenol are expressed in μ M, and velocities are expressed in nmol of product formed/min/mg enzyme. Each data point represents the mean value derived from three determinations.

previous studies have revealed that some divalent metal cations may exert stimulatory or inhibitory effects on the cytosolic STs (10, 11). We were interested in investigating the effects of these divalent metal cations on the sulfation of drug compounds. Enzymatic assays, using

Table 3. Inhibitory/stimulatory effects of divalent metal cations on the human M-form and P-form PSTs.ª

	M-form PST		P-form PST			
	Dobutamine $(5.0 \ \mu M)$	(\pm) Isoproterenol (10 μ M)	Acetaminophen (500 μ M)	Minoxidil (250 µM)		
	(nmol/min/mg protein)					
Control	112 ± 3	126 ± 1	$9.5 \hspace{0.2cm} \pm \hspace{0.2cm} 0.5 \hspace{0.2cm}$	3.1 ± 0.3		
FeCl_2	73.9 ± 1.1	90.7 ± 2.1	$13.1 \hspace{.1in} \pm 0.3 \hspace{.1in}$	$3.3 \hspace{0.2cm} \pm \hspace{0.2cm} 0.2 \hspace{0.2cm}$		
HgCl_2	3.8 ± 0.2	ND^{b}	$1.4 \hspace{0.2cm} \pm \hspace{0.2cm} 0.1 \hspace{0.2cm}$	0.06 ± 0.02		
$CoCl_2$	103 ± 0.2	$119 \hspace{0.1in} \pm 2$	5.3 ± 0.2	$1.2 ext{ }\pm ext{ }0.1 ext{ }$		
$ZnCl_2$	99.6 ± 0.4	65.5 ± 0.4	3.6 ± 0.1	$0.6 \hspace{0.2cm} \pm \hspace{0.2cm} 0.1 \hspace{0.2cm}$		
$Pb(CH_3COO)_2$	107 ± 3	120 ± 1	$10.5 \hspace{0.2cm} \pm \hspace{0.2cm} 0.9 \hspace{0.2cm}$	$3.0 \hspace{0.2cm} \pm \hspace{0.2cm} 0.1 \hspace{0.2cm}$		
$CdCl_2$	$111 \hspace{0.1in} \pm \hspace{0.1in} 2$	122 ± 2	$8.7 \hspace{0.2cm} \pm \hspace{0.2cm} 0.1 \hspace{0.2cm}$	$2.0 \hspace{0.2cm} \pm \hspace{0.2cm} 0.2 \hspace{0.2cm}$		
$MnCl_2$	99.5 ± 1.3	124 ± 3	$12.6 \hspace{0.2cm} \pm \hspace{0.2cm} 0.1 \hspace{0.2cm}$	$3.7 \hspace{0.2cm} \pm \hspace{0.2cm} 0.2 \hspace{0.2cm}$		
$CaCl_2$	114 ± 3	115 ± 3	$10.3 \hspace{0.2cm} \pm \hspace{0.2cm} 0.6 \hspace{0.2cm}$	$2.8 \hspace{0.2cm} \pm \hspace{0.2cm} 0.1 \hspace{0.2cm}$		
$MgCl_2$	113 ± 1	135 ± 2	$10.8 \hspace{0.2cm} \pm \hspace{0.2cm} 0.8 \hspace{0.2cm}$	3.5 ± 0.2		
$CuCl_2$	0.8 ± 0.1	2.4 ± 1.2	0.05 ± 0.02	0.04 ± 0.02		
NaCl	$112 \pm 3 $	$118 \pm 0.3 $	$9.5 \hspace{0.2cm} \pm \hspace{0.2cm} 0.8 \hspace{0.2cm}$	3.0 ± 0.2		

^aData shown represent mean ± SD derived from three determinations. ^bActivity not detected.

acetaminophen and minoxidil (for the P-form PST) or dobutamine and (\pm) isoproterenol (for the M-form PST) as substrates (at the concentrations specified), were carried out in the absence or presence of various divalent metal cations at a concentration of 5 mM. As a control for the counter ion, Cl-, parallel assays in the presence 10 mM NaCl were also performed. Results obtained are compiled in Table 3. The degrees of inhibition or stimulation were evaluated by comparing the activities determined in the presence of metal cations with the activities determined in the absence of metal cations. Both the M-form and Pform PSTs were found to be partially or completely inhibited or stimulated by the majority of the divalent metal cations tested, and both were less sensitive to the effects of physiologically more abundant metal cations such as Mg²⁺ and Ca²⁺. In contrast, they were more sensitive to the detrimental effects of metal cations that may enter the body as environmental contaminants. At a concentration of 5 mM, Cu²⁺ and Hg²⁺ exerted nearly complete inhibition of the activity of the M-form and P-form PSTs toward the four drug compounds tested. These dramatic inhibitory effects imply that the presence of these metal cations may interfere with the metabolism of drugs through sulfation. To what extent the inhibitory effects of these metal cations on the M-form and P-form PSTs may help prolong the presence of drug compounds in vivo, however, remains to be clarified.

Finally, it should be pointed out that, in addition to the differential substrate specificity of the human M-form and P-form PSTs toward drug compounds, recent studies have revealed the genetic polymorphism of the P-form PST (27, 28). It may be important, therefore, to examine the drug-sulfating phenotypes of different P-form PST allozymes. Information of this kind will likely help in understanding the individual differences in drug metabolism and may in the future aid in developing tailored drug regimens that suit individual needs.

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